**Organic practical procedures (Level 3) examiners tips: Read these please!**

• sodium hydrogen carbonate is a base, this can be used to neutralise any excess acid in a reaction

• being a "carbonate" - bubbles will be observed because carbon dioxide gas is given off

• reflux is a way of preventing volatile reactants escaping from the completion mixture

#### don’t be daft!

#### TURN every page of the answer booklet to make sure that you have attempted EVERY question

#### answer every BULLET POINT of every question

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